

Cambridge IGCSE™

BIOLOGY
Paper 6 Alternative to Practical
MARK SCHEME
Maximum Mark: 40

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2021 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

This document consists of 8 printed pages.

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Cambridge IGCSE – Mark Scheme PUBLISHED Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always whole marks (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

© UCLES 2021 Page 2 of 8

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GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.

5 'List rule' guidance

For questions that require *n* responses (e.g. State **two** reasons ...):

- The response should be read as continuous prose, even when numbered answer spaces are provided.
- Any response marked *ignore* in the mark scheme should not count towards *n*.
- Incorrect responses should not be awarded credit but will still count towards n.
- Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
- Non-contradictory responses after the first *n* responses may be ignored even if they include incorrect science.

© UCLES 2021 Page 3 of 8

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Mark scheme abbreviations

		concretes marking no	into
• ;	• •	separates marking po	mus

• I alternative responses for the same marking point

R reject the response
A accept the response
ignore the response
ecf error carried forward
AVP any valid point

ora or reverse argumentAW alternative wording

• underline actual word given must be used by candidate (grammatical variants excepted)

• () the word / phrase in brackets is not required but sets the context

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Question	Answer		Marks	Guidance
1(a)(i)	table drawn with a minimum of three columns and header line; column and row headings; six correct colours recorded;		3	
1(a)(ii)	reducing sugars: Benedict's (solution); protein: biuret (solution); starch: iodine (solution);		3	
1(a)(iii)	starch, protein and (reducing) sugars ;		1	
1(a)(iv)	substance: (reducing) sugars and evidence: positive test / orange, for substance in test-tube /liquid T;		1	
1(b)	idea that some of solution M might have got on to the outside of the tubing / idea of contamination / would give false results as solution M would be in the distilled water / AW;		1	
1(c)	one hazard from:	one precaution from the same row as the hazard:	2	
	(damage to skin from) food testing reagents / chemicals / solutions / liquids	goggles / eye protection / gloves		
	hot water / flame / heat (for Benedict's test)	stand up during practical work / use of a water-bath / (heat-proof) gloves / use of tongs		
		;;		

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October/November 2021

- 5 = -1-5 \cdot -1 \c					
Question	Answer	Marks	Guidance		
1(d)	<pre>independent variable: 1 at least two different pH values;</pre>	6			
	 2,3 variables kept constant, two from: ;; same concentration of amylase same volume of amylase same type of amylase / source of amylase same concentration of starch same volume of starch / size of potato same temperature same time / sample at set time intervals 				
	 method: use of an appropriate reagent to test for starch / reducing sugars; ref. to use of a spotting tile (with iodine); ref. to heating Benedict's solution; use of a buffer; method of maintaining a constant temperature described; 				
	9 repeat the whole investigation at least two times (three trials);				

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Question

2(a)(i)

2(a)(ii)

2(a)(iii)

2(b)(i)

2(b)(ii)

2(b)(iii)

2(b)(iv)

2(c)(i)

temperature;

any two from:

added:

54(%);;;

colour or wavelength of light;

to identify anomalous results;

axes labelled with units;

relevant data quote;

detail 1: five anthers; detail 2: seven petals; stigma labelled;

suitable line;

six correct plots ± half a small square;

Cambridge IGCSE – Mark Scheme October/November 2021 **PUBLISHED** Marks Guidance Answer 1 2 light intensity / distance of lamp/light (from plant) / wattage of light bulb; time, for collecting gas / for equilibration: same / type / species / size / mass / age, of plant; amount / mass / concentration, of sodium hydrogencarbonate / carbon dioxide, 1 MP1 for correct values selected from table (7.8 and 12.0) MP2 for correct calculation MP3 for correct rounding to two significant figures ecf MP2 and MP3 from error in previous

step

4

2

2

5

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correct reading of value from candidate's graph;

size occupies at least half the space available;

outline clear single lines without shading;

even scale that occupies at least half the grid in both directions;

rate / volume, increases (with temperature) and then decreases;

indication that reading has been taken at 17 °C shown on graph;

Question	Answer	Marks	Guidance
2(c)(ii)	line XY 90±1 (mm); 15 mm;;	3	MP1 correct measurement of line XY MP2 correct calculation MP3 correct matching unit ecf MP2 from incorrect measurement

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